

PENANG SANGAM HIGH SCHOOL  
YEAR 11 PHYSICS  
WEEK 17

<b>Strand</b>	ENERGY
<b>Sub Strand</b>	THERMAL ENERGY
<b>Content Learning Outcome</b>	At the end of the lesson students should be able to <ul style="list-style-type: none"> <li>• Apply the law of conservation of energy to solve energy problems</li> </ul>

### THERMAL ENERGY

Heat flows from a region of high temperature to a region of low temperature until all the regions are at the same temperature. Temperature is how hot an object is.

#### *Heat transfer*

- a. **Conduction**- this is the heat transfer in solids without the movement of particles.
- b. **Convection** – this is the heat transfer in liquids and gases with the movement of particles. Convection needs gravity to work therefore it cannot take place in vacuum or space.
- c. **Radiation** – this is the heat transfer through electromagnetic waves. All bodies radiate. This can take place anywhere.
  - i. Dull, black, rough surfaces are good radiators, good absorbers, good emitters but bad reflectors.
  - ii. Shiny, silver and polished surfaces are bad radiators, bad absorbers, bad emitters but good reflectors.

substance	Specific heat capacity in J/kg <sup>0</sup> C
WATER	4200
ALCOHOL	2400
ICE	2100
ALUMINIUM	960
GLASS	670
IRON	440
COPPER	400

When ever there is a temperature change either increase or decrease then the energy can be calculated by the formula

$$E = m c \Delta T$$

M – mass in kg, c is the specific heat capacity,  $\Delta T$  can either be the temperature or change in temperature.

Power is given by the formula

$$P = \frac{\text{ENERGY}}{\text{TIME}}$$

$$P = \frac{\text{WORKDONE}}{\text{TIME}}$$

POWER in watts W, energy in joules J, and time must always be in seconds.

1. 3kg water is at a temperature of  $56^{\circ}\text{C}$ . Find the energy it has.

$$\begin{aligned} E &= m c \Delta T \\ &= 3(4200)(56) \\ &= 705,600\text{J} \end{aligned}$$

2. 2.5kg of copper is at a temperature of  $30^{\circ}\text{C}$ . How much energy is required to get the temperature to  $78^{\circ}\text{C}$ .
3. 453600J of energy is supplied to 3kg of water which is at a temperature of  $20^{\circ}\text{C}$ .
  - i. Find the change in temperature of water
  - ii. Find the new temperature of water
4. A heater is immersed in 2kg of water at  $45^{\circ}\text{C}$ . The temperature becomes  $72^{\circ}\text{C}$  in 5minutes.
  - i. Find the energy given out by the heater
  - ii. Find the heaters power