

**Matching**

Match Column I with Column II by writing the number of the correct choice in the box.

**Column I**

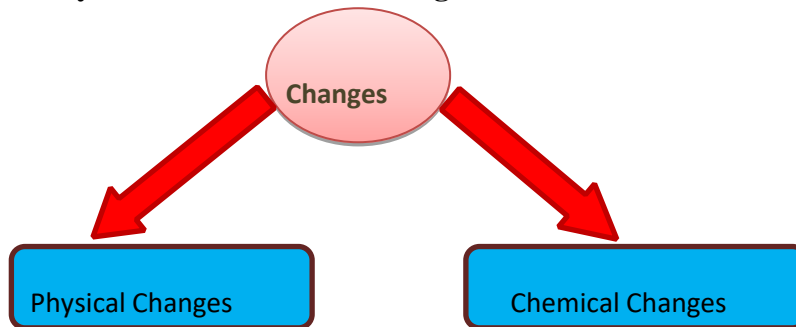
- A. Evaporation
- B. Freezing
- C. Condensation
- D. Melting
- E. Subliming
- F. Disposition


**Column II**

- 1. gas to solid
- 2. solid to gas
- 3. liquid to gas
- 4. solid to liquid
- 5. liquid to solid
- 6. gas to liquid

**Types of Changes**

- Change is an important part of our lives and we cannot ignore change.
- Change can be classified into categories, namely Physical and Chemical change.

**Physical and Chemical Changes.****Part A:****Physical Change**

- Physical Changes are changes affecting the form of a chemical substance.
- They are used to separate mixtures into their component and it can be reversible.
- Some examples are melting ice and boiling water.

**Part B: Chemical Change**

- Chemical changes are changes in which new substances are formed and this cannot be reversed.
- Some examples of chemical changes are:

**QUESTIONS:**

1. What is the difference between Physical change and Chemical change?

---

2. Give an example of a Physical and Chemical change.

---



---

The End